

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

# Methyl 2-(but-3-enyl)-4-hydroxy-1,1-dioxo-2H-1λ<sup>6</sup>,2-benzothiazine-3-carboxylate

Muhammad Nadeem Arshad,<sup>a\*</sup> Islam Ullah Khan,<sup>b</sup>  
Muhammad Zia-ur-Rehman,<sup>c</sup> Muhammad Danish<sup>a</sup> and  
K. Travis Holman<sup>d</sup>

<sup>a</sup>Department of Chemistry, University of Gujrat, Gujrat 50781, Pakistan, <sup>b</sup>Materials Chemistry Laboratory, Department of Chemistry, GC University, Lahore 54000, Pakistan, <sup>c</sup>Applied Chemistry Research Centre PCSIR Laboratories Complex, Lahore 54600, Pakistan, and <sup>d</sup>Department of Chemistry, Georgetown University, 37th and 'O' Streets NW Washington, DC 20057-1227, USA  
Correspondence e-mail: mnachemist@hotmail.com

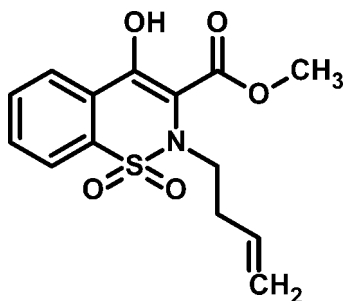
Received 17 May 2012; accepted 19 May 2012

Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.042;  $wR$  factor = 0.109; data-to-parameter ratio = 17.8.

In the title compound,  $\text{C}_{14}\text{H}_{15}\text{NO}_5\text{S}$ , the thiazine ring adopts a sofa conformation and an intramolecular  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bond forms an  $S(6)$  ring. In the crystal, molecules are linked via  $\text{C}-\text{H}\cdots\text{O}$  interactions.

## Related literature

For the synthesis, see: Arshad *et al.* (2011b); Zia-ur-Rehman, *et al.* (2006). For related structures, see: Arshad *et al.* (2011a, 2012). For graph-set notation, see: Bernstein *et al.* (1995).



## Experimental

### Crystal data

$\text{C}_{14}\text{H}_{15}\text{NO}_5\text{S}$   
 $M_r = 309.33$   
Orthorhombic,  $Pbcn$

$a = 25.265$  (8) Å  
 $b = 8.929$  (3) Å  
 $c = 12.584$  (4) Å

$V = 2839.0$  (15) Å<sup>3</sup>  
 $Z = 8$   
Mo  $K\alpha$  radiation

$\mu = 0.25$  mm<sup>-1</sup>  
 $T = 100$  K  
 $0.41 \times 0.36 \times 0.19$  mm

### Data collection

Bruker SMART 1K diffractometer  
Absorption correction: multi-scan  
(*SADABS*; Bruker, 2001)  
 $T_{\min} = 0.905$ ,  $T_{\max} = 0.954$

22868 measured reflections  
3445 independent reflections  
2643 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.081$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.042$   
 $wR(F^2) = 0.109$   
 $S = 1.05$   
3445 reflections  
194 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\max} = 0.33$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.52$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C4}-\text{H4}\cdots\text{O2}^i$	0.95	2.52	3.269 (2)	136
$\text{O1}-\text{H1O}\cdots\text{O4}$	0.84 (3)	1.81 (3)	2.577 (2)	151 (3)

Symmetry code: (i)  $-x + \frac{1}{2}, y + \frac{1}{2}, z$ .

Data collection: *SMART* (Bruker, 2001); cell refinement: *S SAINT* (Bruker, 2001); data reduction: *S SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009) and *X-SEED* (Barbour 2001); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON*.

The authors acknowledge the Higher Education Commission of Pakistan for providing fellowships to MNA (PIN # 042-120607-Ps2-183 and PIN # IRSIP-10-PS-2).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5926).

## References

- Arshad, M. N., Khan, I. U., Zia-ur-Rehman, M., Ahmed, W. & Asiri, A. M. (2012). *Acta Cryst.* **E68**, o1663.  
Arshad, M. N., Khan, I. U., Zia-ur-Rehman, M., Rafique, H. M. & Holman, K. T. (2011a). *Acta Cryst.* **E67**, o1823–o1824.  
Arshad, M. N., Khan, I. U., Zia-ur-Rehman, M. & Shafiq, M. (2011b). *Asian J. Chem.* **23**, o2801–2805.  
Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.  
Bernstein, J., Davis, R. E., Shimon, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.  
Bruker (2001). *SADABS*, *APEX2* and *S SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.  
Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.  
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.  
Zia-ur-Rehman, M., Anwar, J., Ahmad, S. & Siddiqui, H. L. (2006). *Chem. Pharm. Bull.* **54**, 1175–1178.

## supplementary materials

*Acta Cryst.* (2012). E68, o1926 [doi:10.1107/S1600536812022908]

## Methyl 2-(but-3-enyl)-4-hydroxy-1,1-dioxo-2H-1 $\lambda$ <sup>6</sup>,2-benzothiazine-3-carboxylate

Muhammad Nadeem Arshad, Islam Ullah Khan, Muhammad Zia-ur-Rehman, Muhammad Danish and K. Travis Holman

### Comment

Herein we report the crystal structure of the title compound in continuation of our research on the synthesis (Arshad *et al.*, 2011*b*), biological activities (Zia-ur-Rehman *et al.*, 2006) and crystal structures (Arshad *et al.*, 2011*a*, 2012) of thiazine related heterocycles.

The title compound is the *N*-3-butenyl derivative of methyl-4-hydroxy-2*H*-1,2-benzothiazine-3-carboxylate 1,1-dioxide. The planar methyl ester moiety (r.m.s deviation 0.008 Å) is oriented at dihedral angles of 11.39 (10)° and 16.97 (11)° with respect to the thiazine (C1/C6/C7/C8/N1/S1) and benzene (C1/C2/C3/C4/C5/C6) rings. The thiazine ring adopts a sofa conformation and is inclined at 17.27 (12)° with respect to the benzene ring. A six membered (C7/O1/H1O/O4/C9/C8) ring with graph set notation  $S^1_1(6)$  (Bernstein *et al.*, 1995) is formed through an O—H $\cdots$ O intramolecular hydrogen bond. The crystal structure is stabilized by weak C—H $\cdots$ O interactions (Tab. 1).

### Experimental

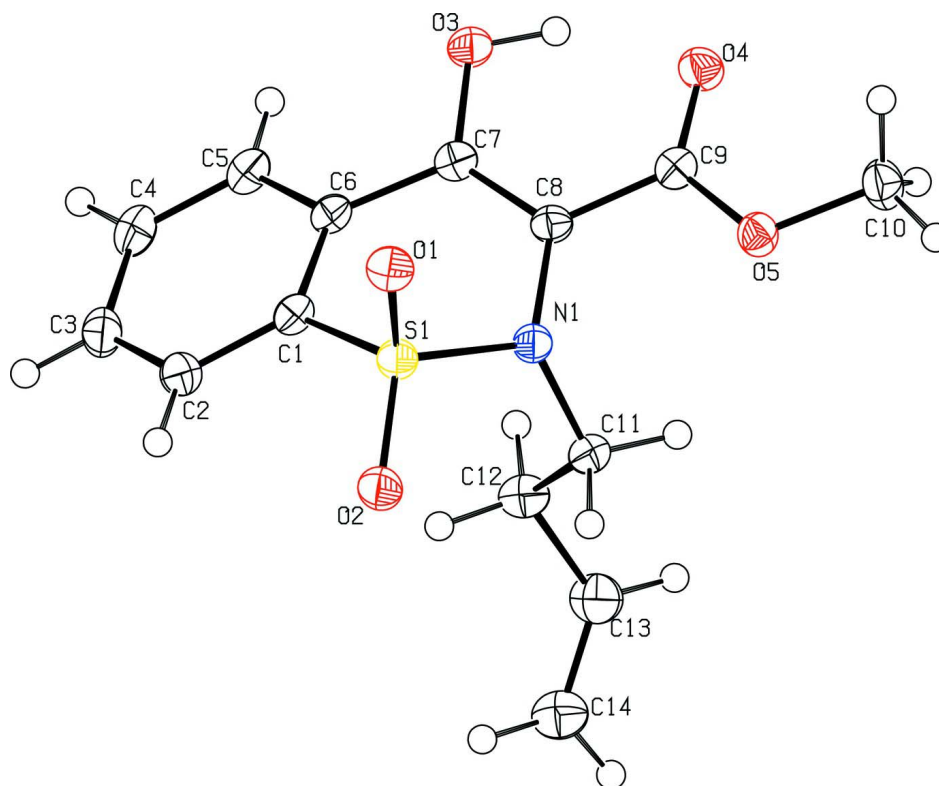
The synthesis of the titled compound has been published (Arshad *et al.*, 2011*a*). The title compound was recrystallized from ethylacetate under slow evaporation of the solvent.

### Refinement

The H-atoms bonded to C were positioned with idealized geometry with C—H ranging from 0.95 Å to = 0.99 Å and refined using a riding model with  $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{C})$  for methyl and  $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$  for the remaining H atoms. The coordinates of the H atom bonded to O were refined with  $U_{\text{iso}}(\text{H}) 1.5U_{\text{eq}}(\text{O})$ .

### Computing details

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINTE* (Bruker, 2001); data reduction: *SAINTE* (Bruker, 2001); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009) and *X-SEED* (Barbour 2001); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON* (Spek, 2009).


**Figure 1**

Molecular structure of the title compound with displacement ellipsoids drawn at the 50% probability.

**Methyl 2-(but-3-enyl)-4-hydroxy-1,1-dioxo-2H-1 $\lambda$ <sup>6</sup>,2-benzothiazine-3-carboxylate**
*Crystal data*

C<sub>14</sub>H<sub>15</sub>NO<sub>5</sub>S

*M<sub>r</sub>* = 309.33

Orthorhombic, *Pbcn*

Hall symbol: -P 2n 2ab

*a* = 25.265 (8) Å

*b* = 8.929 (3) Å

*c* = 12.584 (4) Å

*V* = 2839.0 (15) Å<sup>3</sup>

*Z* = 8

*F*(000) = 1296

*D<sub>x</sub>* = 1.447 Mg m<sup>-3</sup>

Mo *K*α radiation, λ = 0.71073 Å

Cell parameters from 4741 reflections

θ = 2.4–27.3°

μ = 0.25 mm<sup>-1</sup>

*T* = 100 K

Block, colorless

0.41 × 0.36 × 0.19 mm

*Data collection*

Bruker SMART 1K

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

φ and ω scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2001)

*T<sub>min</sub>* = 0.905, *T<sub>max</sub>* = 0.954

22868 measured reflections

3445 independent reflections

2643 reflections with *I* > 2σ(*I*)

*R<sub>int</sub>* = 0.081

θ<sub>max</sub> = 28.3°, θ<sub>min</sub> = 2.4°

*h* = -33→32

*k* = -11→11

*l* = -16→16

Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.042$   
 $wR(F^2) = 0.109$   
 $S = 1.05$   
 3445 reflections  
 194 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H atoms treated by a mixture of independent  
 and constrained refinement  
 $w = 1/[\sigma^2(F_o^2) + (0.0432P)^2 + 1.393P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.33 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.52 \text{ e } \text{\AA}^{-3}$

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.383769 (17)	0.36570 (5)	0.46460 (3)	0.01998 (13)
O1	0.28657 (5)	0.22258 (15)	0.20949 (10)	0.0233 (3)
O2	0.38708 (5)	0.20923 (14)	0.48963 (10)	0.0237 (3)
O3	0.41184 (5)	0.47125 (15)	0.52964 (10)	0.0252 (3)
O4	0.37180 (5)	0.18207 (15)	0.10068 (10)	0.0242 (3)
O5	0.44900 (5)	0.25535 (14)	0.17283 (10)	0.0218 (3)
N1	0.40302 (6)	0.38813 (17)	0.34120 (11)	0.0193 (3)
C1	0.31660 (7)	0.4152 (2)	0.45726 (14)	0.0206 (4)
C2	0.29195 (7)	0.4945 (2)	0.53786 (14)	0.0234 (4)
H2	0.3113	0.5258	0.5987	0.028*
C3	0.23838 (7)	0.5274 (2)	0.52817 (15)	0.0260 (4)
H3	0.2210	0.5839	0.5819	0.031*
C4	0.21024 (7)	0.4782 (2)	0.44037 (15)	0.0256 (4)
H4	0.1734	0.4989	0.4354	0.031*
C5	0.23497 (7)	0.3991 (2)	0.35976 (15)	0.0233 (4)
H5	0.2151	0.3656	0.3003	0.028*
C6	0.28911 (7)	0.36857 (19)	0.36578 (14)	0.0194 (4)
C7	0.31751 (7)	0.29795 (19)	0.27785 (14)	0.0198 (4)
C8	0.37096 (7)	0.3116 (2)	0.26458 (14)	0.0190 (4)
C9	0.39673 (7)	0.2439 (2)	0.17199 (14)	0.0202 (4)
C10	0.47565 (8)	0.1946 (2)	0.08044 (15)	0.0271 (4)
H10A	0.4664	0.0886	0.0725	0.041*
H10B	0.5140	0.2044	0.0894	0.041*
H10C	0.4645	0.2496	0.0169	0.041*
C11	0.42697 (7)	0.5340 (2)	0.30952 (15)	0.0228 (4)

H11A	0.4436	0.5221	0.2387	0.027*
H11B	0.4554	0.5591	0.3606	0.027*
C12	0.38804 (8)	0.6648 (2)	0.30478 (16)	0.0272 (4)
H12A	0.3576	0.6377	0.2591	0.033*
H12B	0.3745	0.6864	0.3770	0.033*
C13	0.41478 (8)	0.8014 (2)	0.26092 (17)	0.0316 (5)
H13	0.4245	0.8001	0.1880	0.038*
C14	0.42573 (8)	0.9223 (2)	0.31568 (17)	0.0306 (5)
H14A	0.4166	0.9276	0.3888	0.037*
H14B	0.4428	1.0045	0.2823	0.037*
H1O	0.3066 (11)	0.199 (3)	0.159 (2)	0.046*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.0183 (2)	0.0232 (2)	0.0185 (2)	-0.00070 (17)	-0.00184 (16)	-0.00094 (17)
O1	0.0193 (6)	0.0285 (7)	0.0221 (7)	-0.0037 (5)	-0.0026 (5)	-0.0032 (5)
O2	0.0240 (7)	0.0246 (7)	0.0227 (7)	0.0010 (5)	-0.0039 (5)	0.0011 (5)
O3	0.0231 (7)	0.0310 (7)	0.0217 (7)	-0.0030 (5)	-0.0018 (5)	-0.0041 (6)
O4	0.0211 (6)	0.0284 (7)	0.0230 (7)	-0.0001 (5)	-0.0031 (5)	-0.0043 (6)
O5	0.0174 (6)	0.0261 (7)	0.0219 (6)	0.0018 (5)	-0.0002 (5)	-0.0021 (5)
N1	0.0171 (7)	0.0230 (8)	0.0178 (7)	-0.0020 (6)	-0.0010 (6)	-0.0018 (6)
C1	0.0194 (8)	0.0202 (9)	0.0221 (9)	-0.0026 (7)	0.0016 (7)	0.0027 (7)
C2	0.0250 (9)	0.0250 (10)	0.0201 (9)	-0.0023 (8)	0.0012 (7)	-0.0004 (7)
C3	0.0264 (9)	0.0256 (10)	0.0260 (10)	0.0012 (8)	0.0076 (7)	0.0014 (8)
C4	0.0193 (9)	0.0278 (10)	0.0297 (10)	0.0015 (7)	0.0036 (7)	0.0062 (8)
C5	0.0192 (9)	0.0267 (10)	0.0241 (9)	-0.0029 (7)	-0.0001 (7)	0.0043 (8)
C6	0.0189 (8)	0.0187 (9)	0.0205 (9)	-0.0026 (7)	0.0001 (7)	0.0028 (7)
C7	0.0200 (8)	0.0194 (9)	0.0198 (9)	-0.0010 (7)	-0.0022 (7)	0.0018 (7)
C8	0.0186 (8)	0.0194 (8)	0.0189 (9)	-0.0013 (7)	-0.0025 (7)	-0.0004 (7)
C9	0.0196 (9)	0.0196 (9)	0.0213 (9)	0.0006 (7)	-0.0008 (7)	0.0027 (7)
C10	0.0241 (9)	0.0319 (11)	0.0253 (10)	0.0034 (8)	0.0040 (8)	-0.0036 (8)
C11	0.0210 (9)	0.0243 (10)	0.0230 (9)	-0.0048 (7)	0.0010 (7)	-0.0020 (8)
C12	0.0281 (10)	0.0249 (10)	0.0286 (10)	-0.0031 (8)	-0.0032 (8)	-0.0006 (8)
C13	0.0398 (12)	0.0307 (11)	0.0242 (10)	-0.0041 (9)	-0.0020 (9)	0.0037 (9)
C14	0.0343 (11)	0.0261 (10)	0.0313 (11)	-0.0013 (8)	-0.0053 (8)	0.0042 (9)

Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

S1—O2	1.4346 (14)	C5—C6	1.397 (2)
S1—O3	1.4356 (13)	C5—H5	0.9500
S1—N1	1.6396 (16)	C6—C7	1.462 (2)
S1—C1	1.7561 (19)	C7—C8	1.366 (2)
O1—C7	1.343 (2)	C8—C9	1.465 (3)
O1—H1O	0.84 (3)	C10—H10A	0.9800
O4—C9	1.227 (2)	C10—H10B	0.9800
O5—C9	1.325 (2)	C10—H10C	0.9800
O5—C10	1.449 (2)	C11—C12	1.528 (3)
N1—C8	1.433 (2)	C11—H11A	0.9900
N1—C11	1.490 (2)	C11—H11B	0.9900

C1—C2	1.385 (3)	C12—C13	1.499 (3)
C1—C6	1.407 (2)	C12—H12A	0.9900
C2—C3	1.391 (3)	C12—H12B	0.9900
C2—H2	0.9500	C13—C14	1.310 (3)
C3—C4	1.386 (3)	C13—H13	0.9500
C3—H3	0.9500	C14—H14A	0.9500
C4—C5	1.385 (3)	C14—H14B	0.9500
C4—H4	0.9500		
O2—S1—O3	119.04 (8)	C8—C7—C6	122.63 (16)
O2—S1—N1	108.04 (8)	C7—C8—N1	121.29 (16)
O3—S1—N1	108.25 (8)	C7—C8—C9	119.96 (16)
O2—S1—C1	108.24 (8)	N1—C8—C9	118.73 (15)
O3—S1—C1	110.00 (8)	O4—C9—O5	123.51 (17)
N1—S1—C1	101.90 (8)	O4—C9—C8	122.61 (16)
C7—O1—H10	104.9 (18)	O5—C9—C8	113.87 (15)
C9—O5—C10	115.33 (14)	O5—C10—H10A	109.5
C8—N1—C11	117.79 (14)	O5—C10—H10B	109.5
C8—N1—S1	114.29 (12)	H10A—C10—H10B	109.5
C11—N1—S1	118.72 (12)	O5—C10—H10C	109.5
C2—C1—C6	121.90 (17)	H10A—C10—H10C	109.5
C2—C1—S1	121.65 (14)	H10B—C10—H10C	109.5
C6—C1—S1	116.46 (13)	N1—C11—C12	114.66 (15)
C1—C2—C3	118.80 (17)	N1—C11—H11A	108.6
C1—C2—H2	120.6	C12—C11—H11A	108.6
C3—C2—H2	120.6	N1—C11—H11B	108.6
C4—C3—C2	120.14 (17)	C12—C11—H11B	108.6
C4—C3—H3	119.9	H11A—C11—H11B	107.6
C2—C3—H3	119.9	C13—C12—C11	110.25 (17)
C5—C4—C3	120.96 (17)	C13—C12—H12A	109.6
C5—C4—H4	119.5	C11—C12—H12A	109.6
C3—C4—H4	119.5	C13—C12—H12B	109.6
C4—C5—C6	120.12 (18)	C11—C12—H12B	109.6
C4—C5—H5	119.9	H12A—C12—H12B	108.1
C6—C5—H5	119.9	C14—C13—C12	124.9 (2)
C5—C6—C1	118.02 (17)	C14—C13—H13	117.6
C5—C6—C7	121.60 (16)	C12—C13—H13	117.6
C1—C6—C7	120.29 (16)	C13—C14—H14A	120.0
O1—C7—C8	122.81 (16)	C13—C14—H14B	120.0
O1—C7—C6	114.53 (15)	H14A—C14—H14B	120.0
O2—S1—N1—C8	61.69 (14)	C5—C6—C7—O1	-20.8 (2)
O3—S1—N1—C8	-168.15 (12)	C1—C6—C7—O1	162.66 (16)
C1—S1—N1—C8	-52.19 (14)	C5—C6—C7—C8	157.21 (18)
O2—S1—N1—C11	-152.15 (12)	C1—C6—C7—C8	-19.3 (3)
O3—S1—N1—C11	-21.99 (15)	O1—C7—C8—N1	-177.64 (15)
C1—S1—N1—C11	93.96 (14)	C6—C7—C8—N1	4.5 (3)
O2—S1—C1—C2	104.25 (16)	O1—C7—C8—C9	0.7 (3)
O3—S1—C1—C2	-27.34 (18)	C6—C7—C8—C9	-177.14 (16)

N1—S1—C1—C2	-142.01 (15)	C11—N1—C8—C7	-110.93 (19)
O2—S1—C1—C6	-75.07 (15)	S1—N1—C8—C7	35.6 (2)
O3—S1—C1—C6	153.34 (13)	C11—N1—C8—C9	70.7 (2)
N1—S1—C1—C6	38.67 (15)	S1—N1—C8—C9	-142.81 (14)
C6—C1—C2—C3	0.6 (3)	C10—O5—C9—O4	2.4 (3)
S1—C1—C2—C3	-178.67 (14)	C10—O5—C9—C8	-177.74 (15)
C1—C2—C3—C4	1.6 (3)	C7—C8—C9—O4	4.8 (3)
C2—C3—C4—C5	-1.7 (3)	N1—C8—C9—O4	-176.84 (16)
C3—C4—C5—C6	-0.3 (3)	C7—C8—C9—O5	-175.05 (16)
C4—C5—C6—C1	2.4 (3)	N1—C8—C9—O5	3.3 (2)
C4—C5—C6—C7	-174.17 (17)	C8—N1—C11—C12	74.1 (2)
C2—C1—C6—C5	-2.6 (3)	S1—N1—C11—C12	-70.88 (19)
S1—C1—C6—C5	176.72 (14)	N1—C11—C12—C13	-173.83 (15)
C2—C1—C6—C7	174.05 (17)	C11—C12—C13—C14	-111.0 (2)
S1—C1—C6—C7	-6.6 (2)		

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C4—H4 $\cdots$ O2 <sup>i</sup>	0.95	2.52	3.269 (2)	136
O1—H1O $\cdots$ O4	0.84 (3)	1.81 (3)	2.577 (2)	151 (3)

Symmetry code: (i)  $-x+1/2, y+1/2, z$ .